Synthesis and Biological Evaluation of a Backbone-Modified Phytoalexin Elicitor

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Abstract: Two suitably protected building blocks (11 and 33) for the preparation of amide-linked heptaglucoside mimetic 2, an analogue of the naturally occurring phytoalexin elicitor 1 a, were readily accessible by glycal chemistry. Sequential elongation of terminal glucuronide 21 with laminaribiosyl hemiaminal 33 and anomeric amine 11 by EDC/HOBt-catalyzed condensation and two-step conversion of the C6-OTr moiety into the corresponding carboxylate function afforded homogeneous carbopeptoid 2 in high overall yield. It was found that replacement of the acetal linkages by the more rigid amide bonds destroys the phytoalexin-elicitor activity.

Introduction

More than ten years ago, Albersheim et al. reported that the branched β -D-glucohexaosyl glucitol **1a** (Figure 1),^[11] isolated from the mycelial walls of *Phytophtora megasperma* (*f.sp. glycinea*), showed phytoalexin-elicitor activity^[2] in soybean. Studies in our laboratory revealed that the synthetically pre-



Figure 1. Structures of the naturally occuring phytoalexin elicitor 1a, methyl heptaglucoside (1b) and carbopeptoid 2.

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Keywords

bioorganic chemistry · carbohydrates · glycosides · peptide bonds · phytochemistry

pared α -methyl 3²,3⁴-di- β -D-glucopyranosylgentiopentaoside (1b) exhibits the same phytoalexin-elicitor activity^[4] as naturally occurring 1a.^[3] Structure – activity studies showed^[5] that the glucosyl units C, C' and E are essential for maximum elicitor activity. However, the biopotency was reduced considerably if the two side-chain glucosyl units C and C' were attached to adjacent backbone glucosyl residues (i.e. the isomer 3³,3⁴-di- β -D-glucopyranosylgentiopentaose).^[6] The outcome of the latter structure–activity studies implies that the biopotency of 1a is more or less governed by a defined spatial alignment of the glucopyranosyl units along the linear sugar backbone. In this context, it was of interest to find out whether restriction of the conformational freedom of the β -(1 \rightarrow 6) backbone linkages

> would impair the phytoalexin-elicitor activity. Thus far, several approaches dealing with the synthesis of carbohydrate mimics containing peptide instead of the natural interglycosidic bonds have been published.^[7] The ease of introduction of an amide function, which is more rigid than a glycosidic bond, prompted us to prepare the amide–saccharide hybrid **2**, the backbone of which consists of a β -(1 \rightarrow 6)-linked glucuronosylamide chain.

Results and Discussion

Prior to the preparation of target compound **2**, efforts were focused on the assembly of the linear β -(1 \rightarrow 6)-amide-linked gentiotetraose^[8] analogue **29**. It was envisaged that the (9 *H*-fluoren-9-ylmethoxy)carbonyl (Fmoc) strategy, as followed earlier for the preparation of carbopeptoids,^[7] could be adopted for the introduction of the requisite β -(1 \rightarrow 6)-amide bonds. To this end, the condensation of the terminal β -glucosylamine **10** with

tions gave fully protected

derivative 12. p-Toluenesulfon-

ic acid (p-TsOH)-mediated de-

tritylation of 12, followed by oxidation^[13] of the resulting

primary alcohol 13 with

PO) and sodium hypochlorite

(NaOCl) under phase-transfer

conditions, afforded the Fmoc-

protected glucuronosylamine

unit 16 in a yield of 81% based

At this stage, the anomeric

amine in 10 was condensed

with the carboxylic acid

derivative 16 by means of the

coupling agent benzotriazol-

phosphate (BOP) in the pres-

ence of N,N-diisopropylethyl-

amine (DIPEA) to give the desired dimer fragment 19(R = NHFmoc) in a yield of

10%. The spectroscopic data

of the major product formed in

the condensation were in full

accordance with the β -(1 \rightarrow 6)-

amide-linked derivative 18, in-

dicating that the intramolecu-

lar amide bond formation

proceeds faster than the corre-

sponding intermolecular pro-

It occurred to us that protec-

tion of the 1-amino function in

11 with the recently devised^[14]

group presented an attractive

alternative. Thus, glucosyl-

amine 11 was converted into

glucosylimide 14 following the

recently reported procedure of Fraser-Reid et al.^[15] Detrityla-

(TCP)

tetrachlorophthaloyl

hexafluoro-

1-vloxytris(dimethylamino)-

phosphonium

on 12

2,2,6,6-tetramethyl-1-piperidinyloxy free radical (TEM-



Scheme 1. i) 3,3-dimethyldioxirane, CH₂Cl₂/acetone, 0 °C, 5 min, quant.; ii) CH₃CN, ZnCl₂, 2 h, 8: 65%, 9: 63%; iii) 0.1 equiv H₂SO₄, THF/H₂O (5:1, v/v), 15 min, quant.; iv) FmocOSu, NaHCO₃, Na₂CO₃, dioxane/H₂O (1:1, v/v), 3 h, 87%; v) a: TCPO, NEt₃, CH₂Cl₂, 1 h; b: Ac₂O, pyr, 12 h, 69%; vi) 4% *p*-TsOH, CH₂Cl₂/MeOH (1:1, v/v), 2 h, 13: 96%, 15: 89%, 23: 94%, 26: 92%; vii) cat. TEMPO, NaOCI, NaHCO₃, NaCI, KBr, Bu₄NCI, CH₂Cl₂/H₂O (1:1, v/v), 30 min, 16: 84%, 17: 81%; viii) BOP, DIPEA, CH₂Cl₂, 1 h; ix) EDC, HOBt, THF, 10h, 25: 84%, 28: 78%; x) ethylenediamine, CH₃CN/EtOH/THF (2:1:1, v/v), 0°C, 10 min, 88%; xi) a: vii; b: NaClO₂, NaH₂PO₄, 2-methyl-2-butene, *t*BuOH/H₂O (1:1, v/v), 14, 24: 84%, 27: 81%; xii) a: cat. KO/Bu, MeOH/CH₂Cl₂ (2:1, v/v), 30 min; b: 3 atm. H₂, Pd/C, CH₂Cl₂/MeOH/H₂O (1:2:1, v/v), 12 h, 28 + 29: 72%.

the Fmoc-protected nonterminal glucuronosylamine unit 16 was examined (see Scheme 1).

The two building units **10** and **16** were readily accessible^[9] starting from commercially available 3,4,6-tri-*O*-benzyl-D-glucal (**4**) and known 3,4-di-*O*-benzyl-6-*O*-trityl-D-glucal (**5**).^[3c] Treatment of **4** with 3,3-dimethyldioxirane (DMD)^[10] afforded the α -1,2-epoxide **6**,^[11] which was converted with acetonitrile in the presence of ZnCl₂ into the α -oxazoline **8** in an overall yield of 65%. Ring-opening of oxazoline **8** with catalytic sulfuric acid^[12] in THF/H₂O furnished β -glucosylamine **10** in a quantitative yield. In a similar way, glucal **5** was transformed into hemiaminal **11**. Treatment of anomeric amine **11** with Fmocoxysuccinimide (FmocOSu) under Schotten – Baumann condi-

tion of 14 and TEMPO-mediated oxidation of the free primary alcohol in 15 gave TCP-protected 1-amino-1-deoxy glucuronic acid 17 in a high yield. Subsequent condensation of glucosylamine 10 with glucuronide 17 under the agency of BOP and DIPEA proceeded smoothly to give dimer 19 in a yield of 73%. It is of interest to note that the coupling efficiency of 10 with 17 to give 19 could be enhanced ($73 \rightarrow 81\%$) by use of the known peptide coupling agent 1-(3-dimethylaminopropyl)-3-ethylcarbodiimide hydrochloride (EDC) in the presence of the nucleophilic catalyst 1-hydroxybenzotriazole (HOBt). It was established that removal of the TCP moiety in 19 with ethylenediamine led to an epimeric mixture of 20 ($\alpha/\beta = 1:2$). Unfortunately, subjection of 19 to different reaction conditions (THF,

cess.

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 $T = 0 - 60 \,^{\circ}\mathrm{C},$ 1-3 equiv ethylenediamine, 5-60 min) not suppress the did anomerisation. Moreover, attempts to convert the α anomer of 20 into the corresponding β -anomer with catalytic amounts (0.1-1.0 equiv) of H_2SO_4 in THF/H₂O met with little success. In order to circumvent anomerisation during deprotection of the TCP group, hemiaminal 11 was now condensed with the known methyl glucuronide **21**^[16] under the influence of EDC/HOBt. Work-up and purification gave the fully protected disaccharide analogue 22, detritylation of which furnished dimer 23 in an overall yield of 84% from 11. Elongation of the amide chain was effected by oxidation of 23 and subsequent condensation of the newly formed carboxylic acid function in 24 with the nonterminal unit 11. TEMPO/NaOCI-mediated oxidation of 23 afforded an inseparable mixture of the desired carboxylate 24, the corresponding aldehyde and its hydrate. Nevertheless, treatment of the crude mixture with NaClO₂ in the presence of the radical scavenger 2-methyl-2-butene resulted in complete oxidation to 24 in a yield of 84%.[17] Condensation of 24 with hemiaminal 11 proceeded smoothly to furnish fully protected trisaccharide mimetic 25. Repetition



of the sequence of reactions described above (i.e., two-step conversion of 25 into 27 followed by coupling with 11) gave tetramer 28. Deprotection of 28 by detritylation, deacetylation and hydrogenolysis yielded the gentiotetraose mimetic 29, the homogeneity and identity of which was firmly established by mass spectrometry as well as 13 C and 1 H NMR spectroscopy (see Figure 2).

The crucial transformation of the C6-OTr into the corresponding carboxylate $(22 \rightarrow 24)$ thus being solved, it is evident that the assembly of target heptamer 2 can be accomplished (see Scheme 2) starting from the bifunctional laminaribiosyl unit 33. The key dimeric hemiaminal 33 was prepared by regioselective

condensation of α -1,2-epoxide **6** with 6-*O*-trityl-D-glucal^[18] (**3**) yielding dimer glucal **30**,^[19] which was subsequently benzylated to afford fully protected glucal **31**. DMD-mediated oxidation of **31** to give α -1,2-oxirane **32**, followed by ZnCl₂-mediated reaction with CH₃CN and subsequent ring-opening of the intermediate oxazoline, led to the exclusive formation of the β -oriented laminaribiosyl hemiaminal **33** in 56% yield. EDC/HOBt-assisted condensation of the anomeric amino function in **33** with the methyl glucuronide **21** gave branched trimer **34** in 79% yield. Removal of the trityl group in **34** with *p*-TsOH and two-step oxidation (TEMPO/NaOCl then NaClO₂) of the CH₂OH function in **35** yielded trimeric carboxylate **36**. Extension of **36** to the

fully protected tetrameric unit 37 was accomplished by EDC/ HOBt-induced coupling with the core building block 11. Tetramer 37 was converted into the corresponding carboxylate 39 by detritylation $(37 \rightarrow 38)$ and oxidation. Coupling of 39 with 1-amino disaccharide 33 furnished the fully protected hexasaccharide 40. Finally, after acidic removal of the trityl group in 40 and oxidation of alcohol 41 to hexasaccharide carboxylate 42, the terminal residue 11 was introduced by the agency of EDC/HOBt to give the heptasaccharide analogue 43 in 77% yield. Treatment of the doubly branched heptamer 43 with p-TsOH followed by Zemplén deacylation of the 2-O-acetyl groups and then hydrogenolysis of the benzyl groups over Pd/C furnished target heptasaccharide mimetic 2 in 70% yield. The structure of 2 was unambiguously confirmed by mass spectrometry and NMR spectroscopy (600 MHz ¹H TOCSY, HH-COSY and CH-COSY). For example, the coupling constants and chemical shifts of the set of four distinct doublets in the ¹HNMR spectrum (see Figure 2, spectrum I) of the tetrameric fragment 29 are



Figure 2. Anomeric regions of the 600 MHz 1 H NMR spectra of tetramer **29** (I) and heptamer **2** (II).

The dynamic behaviour of the amide-linked gentiobiose carbopeptoid **45** (see Figure 3) was simulated by a 10 ns molecular



Figure 3. Structures of α -methyl gentiobioside 44 and carbopeptoid 45.

dynamics run at a constant temperature of 300 K, using the recently developed CHEAT95 force field^[21] for hydrated oligosaccharides. For comparison, the acetal-linked α -methyl gentiobioside **44** (Table 1) was also simulated under the same conditions. The most important difference between gentiobioside **44** and its amide-linked analogue **45** in the simulations is the flexibility of the linkage. The former compound shows many transitions between different conformers, while the latter stays in the same conformation for more than 8 ns. After that, a few transitions to a second conformation can be observed. The average values of the dihedral angles around the linkages of the most significant conformations observed in the simulations are listed in Table 1 and the preferred conformations are shown in Figure 4. It is not excluded that the binding of the flexible acetal-



Figure 4. Stereoviews of the preferred conformations of α -methyl gentiobioside 44 (above) and carbopeptoid 45 (below).

Table 1. Average torsion angles (degrees) and relative occurrence (%) for the most significant conformations of **44** and **45** in the 10 ns MD simulations.

Acetal-linked a-methyl gentiobioside 44

O 5'-C1'-O 6-C 6	C1'-O6-C6-C5	O6-C6-C5-O5	% 55	
- 50	- 170	60		
-40	180	-60	25	
-60	-90	50	5	
- 50	180	170	5	
-40	90	50	5	

O5'-C1'-N6-C6	C1'-N6-C6-C5	N6-C6-C5-O5	¶⁄0	
- 130	180	- 10	95	
60	180	-20	5	

characteristic for the presence of three β - $(1 \rightarrow 6)$ -amide linkages and one α -linked methoxy group (H₁ corresponds to the glucuronide residue at the reducing end of **29** and H₁... to the glucosylamine unit at the nonreducing end). Three additional doublets, which arise from one β - $(1 \rightarrow 6)$ -amide and two β - $(1 \rightarrow 3)$ -glucosidic bonds, are observed in the corresponding spectrum (II) of the heptameric fragment **2**.

Preliminary biological studies^[20] revealed that the amidelinked analogue 2 does not induce phytoalexin accumulation in soybean. linked phytoalexin elicitor **1b** to the plant cell receptor proceeds by an induced-fit mechanism. However, the presence of the amide linkages in the conformationally restrained heptaglucoside analogue **2** may prevent such a mode of binding to the receptor, accounting for its lack of activity.

Conclusion

The results presented in this paper show that elongation of a terminal glucuronide (e.g. 21) with a suitably protected anomeric amine (e.g. 11 or 33) and subsequent two-step conversion of the CH₂OTr function in the growing chain into the corresponding carboxylate presents a convenient approach towards β -(1 \rightarrow 6)-glucuronosylamide carbopeptoids (e.g. 2 and 29). Moreover, it is also apparent that replacement of the β -(1 \rightarrow 6)-acetal linkages in 1b by the more rigid amide bonds completely ruins the phytoalexin-elicitor activity. It may therefore be concluded that incorporation of an amide instead of a natural glycosidic bond into oligosaccharides has a detrimental effect on the molecular geometry and hydrogen-bonding potential.

Experimental Section

¹H and ¹³C NMR spectra were recorded with a Jeol JNM-FX-200 (200/ 50.1 MHz), a Bruker WM-300 (300/75.1 MHz) or a Bruker DMX-600 spectrometer (600/150.3 MHz). Chemical shifts (δ) are given relative to tetramethylsilane as internal standard. Mass spectra were recorded with a Finnigan MAT TSQ70 triple quadropole mass spectrometer. Optical rotations were measured on a Propol automatic polarimeter. Dichloromethane (CH₂Cl₂), pyridine and toluene were heated under reflux with CaH₂ for 3 h, distilled and stored over molecular sieves (4 Å). N,N-Diisopropylethylamine (DIPEA) was subsequently distilled from KOH, ninhydrin and CaH2. Triethylamine was distilled from CaH2. Acetone (Boom Chemicals, c.p.), acetonitrile (Rathburn, HPLC grade), 1,2-dichloroethane (Biosolvent, HPLC grade), N,N-dimethylformamide (DMF, Baker, p.a.), 1,4-dioxane (Baker, p.a.), ethanol (Baker, p.a.) and tetrahydrofuran (THF, Biosolvent, HPLC grade) were stored over molecular sieves (4 Å). Methanol (Rathburn, HPLC grade) was stored over molecular sieves (3 Å). Zinc chloride (Merck, p.a.) was dissolved in THF (1.0 M solution) and stored over molecular sieves (3 Å). Acetic anhydride (Baker, p.a.), benzotriazol-1-yloxytris(dimethylamino)phosphonium hexafluorophosphate (BOP, Richelieu), benzyl bromide (Merck), tert-butanol (Baker, p.a.), 1-(3-dimethylaminopropyl)-3-ethylcarbodiimide hydrochloride (EDC, Acros), ethylenediamine (Acros, p.a.), 1-hydroxybenzotriazole (HOBt, Aldrich), 9-fluorenylmethoxycarbonyloxysuccinimide (FmocOSu, Nova Biochem), 2-methyl-2-butene (Aldrich), Oxone® (Aldrich), palladium on carbon (10%, Acros), potassium tert-butoxide (Aldrich), sodium chlorite (Acros), sodium hydride (Acros, 60% dispersion in mineral oil), sodium hypochlorite (13% active chlorine solution, Acros), tetra-n-butylammonium chloride, tetrachlorophthalic anhydride (TCPO, Acros), 2,2,6,6-tetramethyl-1-piperidinyloxy free radical (TEMPO, Aldrich), p-toluenesulfonic acid monohydrate (Aldrich) and trityl chloride (TrCl, Merck) were used as received. Column chromatography was performed on Baker silica gel (0.063-0.200 mm). Gel permeation chromatography was accomplished on HW-40 column material (Pharmacia). TLC analysis was carried out on prepared plates from Schleicher & Schüll (F1500, LS254) with detection by UV absorption (254 nm) where applicable and charring with 20% H₂SO₄ in MeOH or ammonium molybdate (25 g L⁻¹) and ceric ammonium sulfate (10 gL⁻¹) in 10% aq. H₂SO₄. Reactions were run at ambient temperature, unless otherwise stated. Prior to reactions that required anhydrous conditions, traces of water in the glycosides were removed by coevaporation with 1,2-dichloroethane, pyridine or toluene.

(3 aS,5R,6R,7S,7aR)-6,7-Bis(benzyloxy)-5-(trityloxymethyl)-2-methyl-5Hpyrano[2,3-d]oxazole (9): Under a continuous stream of dry nitrogen, ZnCl₂ (1.0 m solution in THF, 30 mL) was added to a stirred solution of epoxide 7 (11.7 g, 20 mmol) in CH₃CN (100 mL). The reaction mixture was stirred for 2 h, subsequently diluted with EtOAc (400 mL) and washed with sat. aq. NaCl (2 × 100 mL) and aq. NaHCO₃ (1.0 M, 100 mL). The organic phase was dried (MgSO₄) and concentrated in vacuo. Purification of the residue by silica gel chromatography (20–50% EtOAc/light petroleum) afforded oxazoline **9** as a white foam (7.9 g, 12.6 mmol, 63%). ¹H NMR (CDCl₃): δ = 7.52–6.89 (m, 25H, H_{arom}), 5.99 (d, 1H, H1, J_{1,2} = 8.1 Hz), 4.85 (dd, 1H, H2, J_{2,3} = 8.4 Hz), 4.70 (AB, 2H, CH₂ Bn), 4.54 (AB, 2H, CH₂ Bn), 3.81–3.79 (m, 2H, H3/H4), 3.60 (m, 2H, H5/H6), 3.28 (m, 1H, H6'), 2.03 (s, 3H, CH₃); ¹³C{¹H} NMR (CDCl₃): δ = 168.0 (C=N), 143.6 (C_q Tr), 137.6, 137.5 (C_q Bn), 128.5–126.7 (C_{arom}), 93.1 (C1, J_{C,H} = 168.5 Hz), 86.3 (C_q Tr), 80.0, 79.8, 74.5, 71.5 (C2/C3/C4/C5), 73.4, 72.1 (CH₂ Bn), 62.9 (C6), 14.0 (CH₃).

2-O-Acetyl-3,4-di-O-benzyl-6-O-trityl-β-D-glucopyranosylamine (11): To a stirred solution of oxazoline 9 (7.9 g, 12.6 mmol) in THF/H₂O (100 mL, 5:1, v/v) was added aq. H₂SO₄ (1.0 m, 1.26 mL). After 30 min, the reaction mixture was quenched by addition of aq. NaHCO3 (1.0M, 50 mL). The neutralised mixture was extracted with EtOAc (2×200 mL) and the combined organic layers were dried (MgSO₄) and concentrated under reduced pressure. The resulting colourless oil was purified by flash chromatography over silica gel (50% EtOAc/light petroleum) to furnish glucosylamine 11 as a white solid in quantitative yield (8.1 g, 12.6 mmol). $[\alpha]_D = +12.6$ (c = 1, CHCl₃); ¹HNMR (CDCl₃): $\delta = 7.52 - 6.84$ (m, 25H, H_{arom}), 4.90 (dd, 1H, H2, $J_{1,2} = 8.8$ Hz, $J_{2,3} = 9.6$ Hz), 4.67 (AB, 2H, CH₂ Bn), 4.60 (AB, 2H, CH₂ Bn), 4.07 (d, 1 H, H1), 3.86 (dd, 1 H, H3, $J_{3,4} = 9.4$ Hz), 3.69-3.42 (m, 3 H, H4/H5/H6), 3.21 (dd, 1 H, H6', $J_{5,6'}$ = 3.6 Hz, $J_{6,6'}$ = 10.1 Hz), 2.03 (s, 3 H, CH₃ Ac), 1.80 (brs, 2H, NH₂); ¹³C{¹H} NMR (CDCl₃): δ = 170.2 (C=O Ac), 143.5 (C_a Tr), 138.0, 137.5 (C_a Bn), 128.5–126.7 (C_{arom}), 86.1 (C_a Tr), 84.3 (C1, J_{C,H} = 158.2 Hz), 83.3, 78.0, 75.5, 74.2 (C2/C3/C4/C5), 75.0, 74.6 (CH₂ Bn), 62.2 (C6), 20.8 (CH₃ Ac); C₄₁H₄₁NO₆ (643.3): calcd. C 76.49, H 6.42, N 2.18; found C 76.35, H 6.20, N 2.09.

N-(Fluoren-9-ylmethoxycarbonyl)-2-O-acetyl-3,4-di-O-benzyl-6-O-trityl-β-Dglucopyranosylamine (12): To a stirred mixture of glucosylamine 11 (3.22 g, 5.0 mmol), dioxane (25 mL) and H₂O (25 mL) containing NaHCO₃ (1.68 g, 20.0 mmol) and Na2CO3 (1.06 g, 10.0 mmol) was added FmocOSu (2.02 g, 6.0 mmol). After 3 h, the reaction mixture was extracted with EtOAc $(2 \times 100 \text{ mL})$ and the combined organic phases were dried (MgSO₄) and concentrated in vacuo. The resulting pale yellow oil was purified by silica gel chromatography (0-25% EtOAc/light petroleum) to furnish carbamate 12 as a white foam (3.77 g, 4.4 mmol, 87 %). ¹H NMR (CDCl₃): $\delta = 7.87 - 6.93$ (m, 33 H, H_{arom}), 5.81 (d, 1 H, NH, $J_{1, NH} = 6.6$ Hz), 5.07 (dd, 1 H, H1, $J_{1,2} = 6.8$ Hz), 4.76 (AB, 2H, CH₂ Bn), 4.69 (AB, 2H, CH₂ Bn), 4.46 (d, 2H, CH₂ Fmoc), 4.33 (t, 1 H, CH Fmoc), 4.08 (dd, 1 H, H2, $J_{2,3} = 9.6$ Hz), 3.78-3.62 (m, 4H, H3/H4/H5/H6), 3.25 (dd, 1H, H6', $J_{5.6'} = 2.8$ Hz, $J_{6.6'} = 10.1 \text{ Hz}$, 2.07 (s, 3H, CH₃ Ac); ¹³C{¹H} NMR (CDCl₃): $\delta = 171.4$ (C=O Ac), 156.2 (C=O Fmoc), 144.9, 141.7 (C_a Fmoc), 144.2 (C_a Tr), 138.6, 138.1 (C_q Bn), 129.2-120.3 (C_{aron}), 86.8 (C_q Tr), 83.5 (C1), 81.3, 78.2, 76.7, 73.3 (C2/C3/C4/C5), 74.8, 74.5 (CH₂ Bn), 65.4 (CH₂ Fmoc), 62.5 (C6), 47.3 (CH Fmoc), 21.2 (CH₃ Ac); C₅₆H₅₁NO₈ (865.4): calcd. C 77.67, H 5.94, N 1.62; found C 77.70, H 5.98, N 1.51.

N-(Fluoren-9-ylmethoxycarbonyl)-2-O-acetyl-3,4-di-O-benzyl-β-D-glucopyra-

nosylamine (13): A solution of p-TsOH (4%, 4.0 g) in MeOH/CH₂Cl₂ (100 mL, 1:1, v/v) was added to a stirred solution of compound 12 (3.77 g, 4.4 mmol) in CH₂Cl₂ (4 mL). After 2 h, the reaction mixture was neutralised by addition of aq. NaHCO3 (1.0 M, 50 mL) and extracted with EtOAc $(2 \times 100 \text{ mL})$. The combined organic layers were dried (MgSO₄) and concentrated under reduced pressure, and the resulting pale yellow oil was subjected to silica gel chromatography (20-50% EtOAc/light petroleum) to afford alcohol 13 as a white solid (2.60 g, 4.2 mmol, 96%). $^1H\,NMR$ (CDCl_3): δ = 7.78 – 7.30 (m, 18 H, H_{arom}), 6.68 (d, 1 H, NH, $J_{1, NH}$ = 6.4 Hz), 4.85 (dd, 1 H, H1, $J_{1,2} = 7.1$ Hz), 4.82 (AB, 2H, CH₂ Bn), 4.72 (AB, 2H, CH₂ Bn), 4.37 (m, 3H, CH₂ Fmoc, H2), 4.25 (t, 1H, CH Fmoc), 3.90-3.70 (m, 3H, H3/H4/H5), 3.62 (m, 2H, H6/H6'), 1.95 (s, 3H, CH₃ Ac); ¹³C{¹H} NMR $(CDCl_3): \delta = 170.8 (C=O Ac), 155.7 (C=O Fmoc), 143.6, 141.2 (C_a Fmoc),$ 138.1, 137.9 (C_q Bn), 128.3–119.9 (C_{arom}), 83.0 (C1), 80.8, 77.2, 77.1, 72.7 (C2/C3/C4/C5), 75.2, 75.0 (CH₂ Bn), 65.7 (CH₂ Fmoc), 61.0 (C6), 46.8 (CH Fmoc), 20.7 (CH₃ Ac); C₃₇H₃₇NO₈ (623.3): calcd. C 71.25, H 5.98, N 2.25; found C 71.35, H 6.03, N 2.16.

N.N-Tetrachlorophthaloyl-2-O-acetyl-3,4-di-O-benzyl-6-O-trityl-B-D-glucopyranosylamine (14): Tetrachlorophthalic anhydride (1.72 g, 6.0 mmol) was added to a stirred solution of glucosylamine 11 (3.22 g, 5.0 mmol) in CH₂Cl₂ (25 mL) containing triethylamine (0.84 mL, 6.0 mmol). The reaction mixture was stirred for 1 h and concentrated in vacuo, and the resulting slurry was dissolved in pyridine (15 mL). Acetic anhydride (1.42 mL, 15 mmol) was added to the latter solution and the reaction mixture was stirred for 12 h and subsequently concentrated under reduced pressure. The residue was dissolved in EtOAc (100 mL), washed with aq. NaHCO₃ (2 × 50 mL), dried (MgSO₄) and concentrated in vacuo. Traces of pyridine were removed by coevaporation with toluene $(3 \times 50 \text{ mL})$. Purification was accomplished by silica gel chromatography (0-30% EtOAc/light petroleum) to afford glucosylimide 14 as a white foam (3.15 g, 3.5 mmol, 69%). ¹H NMR (CDCl₃): $\delta = 7.51 - 6.94$ (m, 25 H, H_{arom}), 5.88 (dd, 1 H, H2, $J_{1,2} = 9.4$ Hz, $J_{2,3} = 9.2$ Hz), 5.30 (d, 1 H, H1), 4.73 (AB, 2 H, CH₂ Bn), 4.67 (AB, 2 H, CH₂ Bn), 4.07 (dd, 1 H, H3, $J_{3,4} = 9.6 \text{ Hz}$), 3.76 (dd, 1H, H4, $J_{4,5} = 9.2 \text{ Hz}$), 3.62 (dd, 1H, H6, $J_{5,6} = 2.1 \text{ Hz}, J_{6,6'} = 11.0 \text{ Hz}, 3.55 \text{ (m, 1H, H5)}, 3.25 \text{ (dd, 1H, H6', }$ $J_{5,6'} = 4.1 \text{ Hz}$), 1.81 (s, 3H, CH₃ Ac); ¹³C{¹H} NMR (CDCl₃): $\delta = 170.0$ (C=O Ac), 161.6 (C=OTCP), 143.9 (C_q Tr), 140.5, 130.1, 127.3 (C_q TCP), 138.3, 137.8 (C_q Bn), 128.9–127.0 (C_{arom}), 86.8 (C_q Tr), 83.7 (C1), 78.9, 77.8, 77.4, 71.1 (C2/C3/C4/C5), 75.5, 75.2 (CH₂ Bn), 63.0 (C6), 20.7 (CH₃ Ac); C49H39Cl4NO8 (909.1): calcd. C 64.56, H 4.31, N 1.54; found C 64.55, H 4.29, N 1.55.

$\textit{N,N-Tetrachlorophthaloyl-2-O-acetyl-3,4-di-O-benzyl-\beta-D-glucopyranosyl-benz$

amine (15): Compound 14 (3.15 g, 3.5 mmol) was detritylated as described for the preparation of 13 from 12 and subsequently subjected to silica gel chromatography (20–50% EtOAc/light petroleum) to give alcohol 15 as a white solid (2.06 g, 3.1 mmol, 89%). ¹H NMR (CDCl₃): δ =7.45–7.39 (m, 10 H, H_{arom}), 5.91 (dd, 1H, H2, J_{1.2} = 9.2 Hz, J_{2.3} = 7.7 Hz), 5.44 (dd, 1 H, H1, J_{1.3} = 1.5 Hz), 4.96 (AB, 2H, CH₂ Bn), 4.87 (AB, 2H, CH₂ Bn), 4.02–3.84 (m, 4H, H3/H4/H6/H6'), 3.70 (m, 1H, H5), 1.93 (s, 3H, CH₃ Ac); ¹³C{¹H} NMR (CDCl₃): δ =169.9 (C=O Ac), 161.7 (C=O TCP), 140.2, 129.9, 126.9 (C_q TCP), 138.2, 137.9 (C_q Bn), 128.3–127.6 (C_{arom}), 83.4 (C1), 78.6, 78.5, 76.9, 70.7 (C2/C3/C4/C5), 75.2, 75.1 (CH₂ Bn), 61.3 (C6), 20.6 (CH₃ Ac); C₃₀H₂₅Cl₄NO₈ (667.0): calcd. C 53.83, H 3.76, N 2.09; found C 53.70, H 3.84, N 2.09.

N-(Fluoren-9-ylmethoxycarbonyl)-2-O-acetyl-3,4-di-O-benzyl-β-D-glucurono-

pyranosylamine (16): Alcohol 13 (2.60 g, 4.2 mmol) was dissolved in CH₂Cl₂ (20 mL). To this solution was added TEMPO (10 mg, 64 µmol), sat. aq. NaHCO₃ (10 mL), KBr (50 mg, 0.42 mmol) and $(nBu)_4$ NCl (60 mg). The heterogeneous mixture was cooled (0 °C), after which a solution of aq. NaO-Cl (13% active chlorine, 8 mL), sat. aq. NaHCO3 (5 mL) and sat. aq. NaCl (10 mL) was added dropwise over 15 min under vigourous stirring. 15 min after the final addition the reaction mixture was acidified with aq. HCl (0.5 M, 20 mL) and extracted with EtOAc. The combined organic phase was dried (MgSO₄) and concentrated under reduced pressure, and the resulting pale yellow oil was subjected to silica gel chromatography (0-3% MeOH/ CH₂Cl₂) to afford carboxylic acid 16 as a white solid (2.23 g, 3.5 mmol, $84\sqrt[6]{0}$). ¹H NMR (CDCl₃): $\delta = 7.76 - 7.28$ (m, 18H, H_{arom}), 6.91 (d, 1H, NH, $J_{1, \text{NH}} = 5.8 \text{ Hz}$, 4.98 (dd, 1 H, H1, $J_{1, 2} = 8.8 \text{ Hz}$), 4.84 (AB, 2 H, CH₂ Bn), 4.68 (AB, 2H, CH₂ Bn), 4.58 (dd, 1H, H2, $J_{2,3} = 8.1$ Hz), 4.41 (d, 1H, H5, $J_{4.5} = 7.9$ Hz), 4.36-4.22 (m, 3H, CH₂ Fmoc, H4), 4.19 (t, 1H, CH Fmoc), 3.96 (dd, 1 H, H3, $J_{3,4} = 9.2$ Hz), 2.01 (s, 3 H, CH₃ Ac); ¹³C{¹H} NMR $(CDCl_3): \delta = 171.4, 170.7 (C=O Ac, C6), 155.6 (C=O Fmoc), 143.4, 140.9$ (C_q Fmoc), 137.8, 137.4 (C_q Bn), 128.2–119.7 (C_{arom}), 81.9 (C1), 80.6, 79.5, 75.7, 71.9 (C2/C3/C4/C5), 75.2, 74.7 (CH, Bn), 67.5 (CH, Fmoc), 46.5 (CH Fmoc), 20.6 (CH₃ Ac); MS (ESI): $m/z = 638 (M + H^+)$, $655 (M + NH_4^+)$, 660 $(M + Na^+)$; C₃₇H₃₅NO₉ (637.2): calcd. C 69.69, H 5.53, N 2.20; found C 69.52, H 5.57, N 2.18.

$\textit{N,N-Tetrachlorophthaloyl-2-O-acetyl-3,4-di-O-benzyl-\beta-D-glucuronopyrano-benzyl-ben$

sylamine (17): Alcohol 15 (2.06 g, 3.1 mmol) was oxidised with TEMPO/ NaOCl as described for the synthesis of 16 from 13 and the resulting pale yellow oil was subjected to silica gel chromatography (0-3% MeOH/ CH₂Cl₂) to afford carboxylic acid 17 as a white solid (1.70 g, 2.5 mmol, 81%). ¹H NMR (CDCl₃): δ = 7.42–7.37 (m, 10 H, H_{arom}), 6.03 (dd, 1 H, H2, $J_{1,2}$ = 9.4 Hz, $J_{2,3}$ = 7.5 Hz), 5.60 (d, 1 H, H1), 4.96–4.77 (m, 4H, 2 × CH₂ Bn), 4.36 (d, 1 H, H5, $J_{4,5}$ = 7.9 Hz), 4.19 (dd, 1 H, H4, $J_{3,4}$ = 9.0 Hz), 3.94 (dd, 1 H, H3), 1.95 (s, 3 H, CH₃ Ac); ¹³C{¹H} NMR (CDCl₃): δ = 169.6, 168.9 (C6, C=O Ac), 161.3 (C=O TCP), 140.1, 129.7, 126.6 (C₄ TCP), 137.6, 137.1 (C_q Bn), 127.9–127.3 (C_{arom}), 82.4 (C1), 78.6, 78.0, 76.1, 69.5 (C2/C3/C4/C5), 77.3, 76.6 (CH₂ Bn), 19.9 (CH₃ Ac); MS (ESI): m/z = 682 ($M + H^+$), 699 ($M + NH_4^+$); C₃₀H₂₃Cl₄NO₉ (681.0): calcd. C 52.73, H 3.39, N 2.05; found C 52.70, H 3.42, N 2.05.

N-Fluoren-9-ylmethoxycarbonyl-2-O-acetyl-3,4-di-O-benzyl-β-D-glucuronopyranosylamide (18): BOP (0.243 g, 0.55 mmol) was added to a stirred solution of glucosylamine 10 (0.246 g, 0.5 mmol), glucuronide 16 (0.319 g, 0.5 mmol) and DIPEA (0.26 mL, 1.5 mmol) in CH₂Cl₂ (3 mL). Stirring was continued for 1 h, after which the reaction mixture was diluted with EtOAc (50 mL) and washed with aq. phosphate buffer (pH = 7, 0.1 M, 2×10 mL), dried (MgSO₄) and concentrated in vacuo. Purification of the residue by silica gel chromatography (0-30% EtOAc/light petroleum) afforded glucuronosylamide 18 as a white foam (0.238 g, 0.39 mmol, 77%). When the reaction was carried out in the absence of 10, the same yield of cyclisation product 18 was obtained. ¹H NMR (CDCl₃): $\delta = 7.78 - 7.07$ (m, 18 H, H_{arom}), 5.69 (t, 1 H, H2, $J_{1,2} = J_{2,3} = 1.7$ Hz), 4.94 (d, 1 H, H1), 4.58 (d, 1 H, H5, $J_{4,5} = 2.2$ Hz), 4.53-4.32 (m, 6H, 2×CH₂ Bn, CH₂ Fmoc), 4.18 (t, 1H, CH Fmoc), 3.72 $(dd, 1H, H4, J_{3,4} = 1.7 Hz), 3.65 (t, 1H, H3), 2.14 (s, 3H, CH_3 Ac); {}^{13}C{}^{1}H{}$ NMR (CDCl₃): δ = 169.8, 166.8 (C=O Ac, C6), 149.2 (C=O Fmoc), 142.8, 141.1 (C_a Fmoc), 136.8, 136.7 (C_a Bn), 128.4–119.8 (C_{arom}), 86.5 (C1), 76.6, 73.6, 71.8, 65.3 (C2/C3/C4/C5), 71.9, 71.4 (CH₂ Bn), 68.8 (CH₂ Fmoc), 46.3 (CH Fmoc), 20.8 (CH₃ Ac); MS (ESI): m/z = 620 ($M + H^+$), 637 $(M + NH_4^+)$, 642 $(M + Na^+)$, 658 $(M + K^+)$; $C_{37}H_{33}NO_8$ (619.3): calcd. C 71.79, H 5.33, N 2.26; found C 71.76, H 5.35, N 2.19.

6-*N*-(2-*O*-Acetyl-3,4,6-tri-*O*-benzyl-β-D-glucopyranosyl)-1-*N*,*N*-tetrachlorophthaloyl-2-*O*-acetyl-3,4-di-*O*-benzyl-β-D-amidoglucuronopyranosylamine (19):

Method A: BOP (0.243 g, 0.55 mmol) was added to a stirred solution of glucosylamine **10** (0.246 g, 0.5 mmol), glucuronide **17** (0.340 g, 0.5 mmol) and DIPEA (0.26 mL, 1.5 mmol) in CH₂Cl₂ (3 mL). Stirring was continued for 1 h, after which the reaction mixture was diluted with EtOAc (50 mL) and washed with aq. phosphate buffer (pH = 7, 0.1 M, 2×10 mL), dried (MgSO₄) and concentrated in vacuo. Purification of the residue was effected by silica gel chromatography (10–30% EtOAc/light petroleum) to yield dimer **19** as a white foam (0.422 g, 0.37 mmol, 73%).

Method B: EDC (0.105 g, 0.55 mmol) was added to a solution of glucosylamine 10 (0.246 g, 0.5 mmol), glucuronide 17 (0.340 g, 0.5 mmol) and HOBt (0.068 g, 0.5 mmol) in THF (3 mL) and the solution was stirred for 10 h. Subsequently, the reaction mixture was diluted with EtOAc (50 mL), washed with aq. NaHCO₃ (1.0 M, 3×10 mL), dried (MgSO₄) and concentrated in vacuo. Purification of the residue was accomplished by silica gel chromatography (10-30% EtOAc/light petroleum) to give dimer 19 as a white foam (0.468 g, 0.41 mmol, 81%). ¹H NMR (CDCl₃): $\delta = 7.38 - 7.10 \text{ (m}, 25 \text{ H},$ H_{arom}), 7.14 (d, 1 H, NH, $J_{NH, 1'} = 7.4$ Hz), 5.95 (dd, 1 H, H2, $J_{1, 2} = 9.4$ Hz, $J_{2,3} = 8.5 \text{ Hz}$, 5.37 (d, 1 H, H1), 5.07 (dd, 1 H, H1', $J_{1',2'} = 9.2 \text{ Hz}$), 4.85 (ÅB, 2 H, CH₂ Bn), 4.76 (dd, 1 H, H2', $J_{2', 3'} = 10.6$ Hz), 4.72 (AB, 2 H, CH₂ Bn), 4.63 (AB, 2H, CH, Bn), 4.56 (AB, 2H, CH, Bn), 4.45 (AB, 2H, CH, Bn), 4.07 (d, 1 H, H5, $J_{4,5} = 8.7$ Hz), 3.85 (dd, 1 H, H4, $J_{3,4} = 8.3$ Hz), 3.75 (dd, 1H, H3), 3.73-3.68 (m, 3H, H3'/H4'/H5'), 3.67 (dd, 1H, H6'A, $J_{5', 6'A} = 2.9 \text{ Hz}, J_{6'A, 6'B} = 9.2 \text{ Hz}), 3.55 \text{ (dd, 1 H, H6'B, } J_{5', 6'B} = 4.6 \text{ Hz}),$ 1.87, 1.80 (2×s, 2×3H, 2CH₃ Ac); ${}^{13}C{}^{1}H{}$ NMR (CDCl₃): $\delta = 170.6$, 169.4, 167.8 (C6, 2C=O Ac), 161.5 (C=O TCP), 140.3, 129.9, 126.9 (C_a TCP), 138.0, 137.7, 137.7, 137.5, 137.4 (C_q Bn), 128.1–127.5 (C_{arom}), 82.9, 82.4 (C1/C1'), 78.6, 77.8, 77.5, 77.3, 76.4, 76.1, 72.4, 69.2 (C2-C5, C2'-C5'), 75.2, 75.1, 74.9, 74.8, 73.3 (CH₂ Bn), 67.9 (C6'), 20.5, 20.4 (CH₃ Ac); MS (ESI): $m/z = 1155 (M + H^+)$, 1172 $(M + NH_4^+)$; $C_{59}H_{54}Cl_4N_2O_{14}$ (1154.2): calcd. C 61.25, H 4.70, N 2.42; found C 61.24, H 4.76, N 2.43.

6-*N*-(2-*O*-Acetyl-3,4,6-tri-*O*-benzyl-β-D-glucopyranosyl)-2-*O*-acetyl-3,4-di-*O*-benzyl-α,β-D-amidoglucuronopyranosylamine (**20**): To a stirred solution of dimer **19** (0.578 g, 0.50 mmol) in CH₃CN/EtOH/THF (2:1:1, v/v/v, 4 mL) was added ethylenediamine (0.13 mL, 2.0 mmol). After 10 min, the reaction mixture was concentrated under reduced pressure at room temperature, suspended in CH₂Cl₂ (25 mL) and filtered through a bed of silica (2 g). The filtrate was concentrated in vacuo to afford anomeric mixture **2** as a white solid (0.391 g, 0.44 mmol, 88 %, α/β = 1:2). α-Anomer: ¹³C{¹H} NMR (CD-Cl₃): δ = 172.0, 170.5, 169.3 (C6, 2C=O Ac), 138.1, 137.8, 137.7, 137.6, 137.5, (C_q Bn), 128.3–127.6 (C_{arom}), 83.0, 82.1 (C1/C1'), 78.3, 77.4, 76.5, 75.1, 73.5, 72.5, 72.4, 71.0 (C2–C5, C2'–C5'), 75.0, 74.8, 73.5, 72.6, 72.1 (CH₂ Bn), 67.8 (C6'), 20.9, 20.3 (CH₃ Ac); β-Anomer: ¹³C{¹H} NMR (CDCl₃): δ = 171.0,

170.2, 169.4 (C6, 2C=O Ac), 138.1, 137.8, 137.7, 137.6, 137.5 (C_q Bn), 128.3 - 127.6 (C_{arom}), 84.1, 83.1 (C1/C1'), 79.6, 77.9, 77.4, 76.5, 76.4, 75.2, 74.1, 72.9 (C2–C5, C2'–C5'), 75.3, 75.0, 74.8, 74.6, 73.5 (CH₂ Bn), 68.1 (C6'), 20.9, 20.8 (CH₃ Ac); $C_{51}H_{56}N_2O_{12}$ (888.4): calcd. C 68.90, H 6.35, N 3.15; found C 68.94, H 6.42, N 3.08.

General procedure for amide bond formation: EDC (0.210 g, 1.1 mmol) was added to a stirred solution of HOBt (0.135 g, 1.0 mmol), the appropriate glucosylamine (1.0 mmol) and glucuronic acid derivative (1.0 mmol) in THF (5 mL). When TLC analysis (40% EtOAc/light petroleum) indicated complete conversion of the starting materials (10 h), the reaction mixture was diluted with EtOAc (50 mL), washed with aq. NaHCO₃ (1.0 m, 3×10 mL), dried (MgSO₄) and concentrated in vacuo. Purification of the residue was accomplished by silica gel chromatography (10–40% EtOAc/light petroleum) to afford the corresponding amide-linked oligomer as a white foam.

Methyl 6-N-(2-O-acetyl-3,4-di-O-benzyl-6-O-trityl-B-D-glucopyranosyl)-2,3,4tri-O-benzyl-α-D-amidoglucuronopyranoside (22): Yield: 0.982 g, 0.89 mmol, 89%; ¹H NMR (CDCl₃): δ = 7.47–6.86 (m, 40 H, H_{arom}), 6.97 (d, 1 H, NH, $J_{\rm NH, 1'} = 9.4 \text{ Hz}$), 5.23 (dd, 1 H, H1', $J_{1', 2'} = 9.6 \text{ Hz}$), 4.97 (dd, 1 H, H2', $J_{2', 3'} = 8.9$ Hz), 4.91 (d, 1 H, H1, $J_{1, 2} = 3.8$ Hz), 4.85–4.62 (m, 10 H, 5 CH₂ Bn), 4.39 (d, 1 H, H5, $J_{4,5} = 10.5$ Hz), 4.18 (dd, 1 H, H4, $J_{3,4} = 8.7$ Hz), 4.14 (dd, 1 H, H3, $J_{2,3} = 9.4$ Hz), 3.97 (dd, 1 H, H3', $J_{3',4'} = 9.4$ Hz), 3.72 (t, 1 H, H4', $J_{4',5'} = 9.4$ Hz), 3.68 (dd, 1H, H6'A, $J_{5',6'A} = 3.0$ Hz, $J_{6'A,6'B} =$ 11.9 Hz), 3.62 3.50 (m, 2H, H2/H5'), 3.44 (s, 3H, OMe), 3.20 (dd, 1H, H6'B, $J_{5',6'B} = 4.1$ Hz), 1.87 (s, 3H, CH₃ Ac); ¹³C{¹H} NMR (CDCl₃): $\delta = 170.2$, 169.0 (C6, C=O Ac), 143.3 (C_g Tr), 138.1, 137.7, 137.5, 137.3, 137.2 (C_q Bn), 128.2-126.5 (C_{arom}), 98.0 (C1), 85.9 (C_q Tr), 82.7 (C1'), 80.8, 79.8, 79.0, 77.6, 77.4, 76.2, 72.7, 70.2 (C2-C5, C2'-C5'), 75.4, 75.0, 74.6, 74.5, 73.0 (CH₂ Bn), 61.6 (C6'), 55.2 (OMe), 20.2 (CH₃ Ac); MS (ESI): $m/z = 1104 (M + H^+), 1121 (M + NH_4^+), 1126 (M + Na^+); C_{69}H_{69}NO_{12}$ (1103.5): caled. C 75.05, H 6.30, N 1.27; found C 75.10, H 6.40, N 1.20.

General procedure for detritylation: A solution of *p*-TsOH (4%, 1.0 g) in McOH/CH₂Cl₂ (25 mL, 1:1, v/v) was added to a stirred solution of the 6-O-tritylated oligomer (1.0 mmol) in CH₂Cl₂ (2 mL). After 2 h, the reaction mixture was neutralised by addition of aq. NaHCO₃ (1.0 M, 50 mL) and extracted with EtOAc (2 × 100 mL). The combined organic layers were concentrated under reduced pressure and the resulting pale yellow oil was subjected to silica gel chromatography (0–2% McOH/CH₂Cl₂) to furnish the corresponding primary alcohol as a white solid.

Methyl 6-*N*-(2-*O*-acetyl-3,4-di-*O*-benzyl-β-D-glucopyranosyl)-2,3,4-tri-*O*-benzyl-α-D-amidoglucuronopyranoside (23): Yield: 0.720 g, 0.84 mmol, 94%; ¹H NMR (CDCl₃): $\delta = 7.33 - 7.25$ (m, 25H, H_{arom}), 6.88 (d, 1H, NH, $J_{\rm NH,1'} = 9.6$ Hz), 5.17 (dd, 1H, H1', $J_{1',2'} = 9.2$ Hz), 4.89 (dd, 1H, H2', $J_{2',3'} = 10.7$ Hz), 4.82 (d, 1 H, H1, $J_{1,2} = 4.5$ Hz), 4.79 - 4.60 (m, 10H, 5CH₂ Bn), 4.03 (d, 1H, H5, $J_{4,5} = 9.9$ Hz), 4.01 (dd, 1H, H4, $J_{3,4} = 6.7$ Hz), 3.80 (dd, 1 H, H3, $J_{2,3} = 8.8$ Hz), 3.75 (dd, 1H, H6'A, $J_{5',6'A} = 3.1$ Hz, $J_{6'A,6'B} = 7.1$ Hz), 3.70 - 3.45 (m, 5H, H2/H3'/H4'/H5'/H6'B), 3.37 (s, 3H, OMe), 1.88 (s, 3H, CH₃ Ac), 1.78 (brs, 1H, OH); ¹³C[¹H] NMR (CDCl₃): $\delta = 171.0$, 170.1 (C6, C=O Ac), 138.7, 138.6, 138.3, 138.1, 138.0 (C₄ Bn), 128.5 - 127.7 (C_{arom}), 98.7 (C1), 83.1 (C1'), 81.3, 80.0, 79.5, 78.0, 77.6, 77.4, 77.3, 70.8 (C2 · C5, C2' - C5'), 75.9, 75.4, 75.1, 73.5, 73.4 (CH₂ Bn), 61.1 (C6'), 55.7 (OMe), 20.8 (CH₃ Ac); C_{50} H₅₅NO₁₂ (861.4): calcd. C 69.67, H 6.43, N 1.62; found C 69.50. H 6.46, N 1.63.

General procedure for oxidation of primary alcohols to carboxylic acids: To a solution of the primary alcohol (1.0 mmol) in CH_2Cl_2 (20 mL) was added TEMPO (2 mg, 13 µmol), sat. aq. NaHCO₃ (2 mL), KBr (10 mg, 0.08 mmol) and $(nBu)_4$ NCl (15 mg). This heterogeneous mixture was cooled (0 °C), after which a solution of aq. NaOCl (13% active chlorine, 2 mL), sat. aq. NaHCO₃ (1 mL) and sat. aq. NaCl (2 mL) was added dropwise over 15 min under vigorous stirring. 15 min after the last addition the reaction mixture was acidified with aq. HCl (0.5M, 10 mL) and extracted with EtOAc (2 × 50 mL). The combined organic phase was dried (MgSO₄) and concentrated under reduced pressure, and the resulting pale yellow oil was dissolved in *t*BuOH (20 mL), 2-methyl-2-butene (5 mL) and H₂O (20 mL). To this solution was subsequently added NaH₂PO₄·H₂O (2.0 g, 14 mmol) and NaClO₂ (2.0 g, 22 mmol). The combined organic layers were dried

 $(MgSO_4)$ and concentrated under reduced pressure, and the resulting colourless oil was purified by silica gel chromatography $(0-3\% MeOH/CH_2CI_2)$ to afford the corresponding carboxylic acid derivative as a white solid.

Methyl 6-*N*-(2-*O*-acetyl-3,4-di-*O*-benzyl-β-D-glucuronopyranosyl)-2,3,4-tri-*O*-benzyl-α-D-amidoglucuronopyranoside (24): Yield: 0.544 g, 0.62 mmol. 84%; ¹H NMR (CDCl₃): $\delta = 7.36 - 7.23$ (m, 25H, H_{arom}), 7.08 (d, 1H, NH, $J_{\rm NH,1'} = 9.3$ Hz), 5.13 (t, 1H, H1', $J_{1',2'} = 9.3$ Hz), 4.90–4.65 (m, 10H, 5CH₂ Bn), 4.77 (dd, 1H, H2', $J_{2',3'} = 9.5$ Hz), 4.58 (d, 1H, H1, $J_{1,2} = 4.6$ Hz), 4.10 (d, 1H, H5', $J_{4',5'} = 9.2$ Hz), 4.03 (d, 1H, H5, $J_{4,5} = 10.0$ Hz), 3.97 (dd, 1H, H4, $J_{3,4} = 9.4$ Hz), 3.78 (dd, 1H, H4', $J_{3',4'} = 8.7$ Hz), 3.76 (dd, 1H, H3'), 3.49 (dd, 1H, H2, $J_{2,3} = 9.2$ Hz), 3.46 (dd, 1H, H3'), 3.49 (dd, 1H, H2, $J_{2,3} = 9.2$ Hz), 3.46 (dd, 1H, H3'), 3.49 (s, 3H, OMe), 1.86 (s, 3H, CH₃ Ac); ¹³C{¹H} NMR (CDCl₃): $\delta = 171.1$, 170.7, 170.3 (C6, C6', C=O Ac), 138.0, 137.4, 137.4, 137.1, 137.0 (C₄ Bn), 128.0 - 127.4 (C_{arom}), 98.1 (C1), 81.8 (C1'), 80.8, 79.4, 78.9, 78.8, 77.5, 75.6, 72.0, 70.1 (C2–C5, C2'–C5'), 75.4, 74.9, 74.8, 73.2, 73.1 (CH₂ Bn), 55.1 (OMe), 20.2 (CH₃ Ac); C₅₀H₅₃NO₁₃ (875.4): caled. C 68.56, H 6.10, N 1.60; found C 68.54, H 6.25, N 1.54.

Methyl 6-*N*-[6-*N*-(2-*O*-acetyl-3,4-di-*O*-benzyl-6-*O*-trityl-β-D-glucopyranosyl)-2-*O*-acetyl-3,4-di-*O*-benzyl-β-D-amidoglucuronopyranosyl]-2,3,4-tri-*O*-benzylα-D-amidoglucuronopyranoside (25): Yield: 0.782 g, 0.52 mmol, 84%; $^{13}C{^{11}H}$ NMR (CDCl₃): $\delta = 170.5$, 170.3, 170.0, 168.8 (C6/C6', 2 C=O Ac), 143.4 (C_q Tr), 138.0, 137.7, 137.5, 137.4, 137.3, 137.2, 136.9 (C_q Bn), 128.3-126.7 (C_{arom}), 97.8 (C1), 86.0 (C_q Tr), 82.8, 81.7 (C1'/C1"), 80.7, 80.1, 79.4, 79.0, 77.4, 77.1, 76.9, 76.3, 76.1, 72.7, 71.9, 70.1 (C2–C5, C2'–C5', C2"– C5"), 75.6, 75.2, 75.0, 74.7, 74.6, 73.2, 73.1 (CH₂ Bn), 61.7 (C6"), 54.6 (OMe). 20.3, 20.2 (CH₃ Ac); MS (ESI): m/z = 1502 (M+H⁺), 1519 (M+NH₄⁺), 1524 (M+Na⁺); C₉₁H₉₂N₂O₁₈ (1500.6): calcd. C 72.78, H 6.17, N 1.87; found C 72.80, H 6.20, N 1.86.

Methyl 6-*N*-[6-*N*-(2-*O*-acetyl-3,4-di-*O*-benzyl-β-D-glucopyranosyl)-2-*O*-acetyl-3,4-di-*O*-benzyl-β-D-amidoglucuronopyranosyl]-2,3,4-tri-*O*-benzyl-α-D-amidoglucuronopyranoside (26): Yield: 0.602 g, 0.48 mmol, 92%; $^{13}C{^{1}H}$ NMR (CDCl₃): $\delta = 170.8$, 170.4, 169.7, 168.4 (C6/C6', 2C=O Ac). 138.3, 138.0, 137.7, 137.6, 137.6, 137.5, 137.4 (C_q Bn), 128.2-127.4 (C_{arom}), 98.1 (C1), 82.8, 81.4 (C1'/C1''), 80.9, 80.0, 79.2, 79.1, 78.6, 77.8, 77.2, 77.1, 75.7, 72.7, 72.4, 70.3 (C2-C5, C2'-C5', C2''-C5''), 75.6, 74.9, 74.8, 74.6, 74.4, 73.2, 73.2 (CH₂ Bn), 60.9 (C6''), 55.5 (OMe). 20.5, 20.4 (CH₃ Ac); C₇₂H₇₈N₂O₁₈ (1258.5): calcd. C 68.67, H 6.24, N 2.22; found C 68.69, H 6.20, N 2.19.

Methyl 6-*N*-[6-*N*-(2-*O*-acetyl-3,4-di-*O*-benzyl-β-D-glucuronopyranosyl)-2-*O*-acetyl-3,4-di-*O*-benzyl-β-D-amidoglucuronopyranosyl]-2,3,4-tri-*O*-benzyl-α-D-amidoglucuronopyranoside (27): Yield: 0.391 g, 0.31 mmol, 81%; $^{13}C{^{1}H}$ NMR (CDCl₃): $\delta = 171.0$, 170.4, 169.9, 169.8, 169.2 (C6/C6', C6'', 2C=O Ac), 138.3, 137.7, 137.6, 137.5, 137.4, 137.3, 137.1 (C_q Bn), 128.2–127.5 (C_{arom}), 98.1 (C1), 81.8, 81.3 (C1'/C1"), 81.0, 80.2, 79.1, 79.1, 78.2, 77.9, 76.6, 75.5, 75.2, 72.6, 72.0, 70.3 (C2–C5, C2'–C5', C2"–C5"), 75.7, 75.0, 75.0, 74.9, 74.5, 74.3, 73.3 (CH₂ Bn), 55.6 (OMe), 20.6, 20.3 (CH₃ Ac); C₇₂H₇₆N₂O₁₉ (1272.5): calcd. C 67.91, H 6.02, N 2.20; found C 68.00, H 6.20, N 2.11.

General procedure for deblocking of the amide-linked oligomers: The fully protected oligomer was detritylated as described in the general procedure to give the corresponding primary alcohol. To a stirred solution of the appropriate alcohol (0.10 mmol) in MeOH/CH₂Cl₂ (5 mL, 2:1, v/v) was added KOtBu (11 mg, 0.10 mmol). After 30 min, the reaction was terminated by addition of DOWEX-H⁺ (100 mg). The ion-exchange resin was filtered

off and the filtrate was concentrated in vacuo. The resulting colourless oil was dissolved in $CH_2Cl_2/MeOH/H_2O$ (1:2:1, v/v/v, 4 mL), after which palladium on carbon (10%, 100 mg) was added. The heterogeneous mixture was hydrogenated at clevated pressure (3 atm) in a Parr apparatus for 12 h and subsequently filtered. The filtrate was concentrated under reduced pressure and subjected to Pharmacia HW-40 gel filtration (eluent: H_2O) and lyophilised to afford the corresponding unprotected amide-linked oligomer.

Methyl 6-N-[6-N-[β-D-glucopyranosyl]-β-D-amidoglucuronopyranosyl]-β-D-am

1,5-Anhydro-3-*O***-**(**3,4,6-tri-***O***-benzyl-** β -D-glucopyranosyl)-6-*O*-trityl-2-deoxy-D-*arabino*-hex-1-enitol (**30**): Under a continuous stream of dry nitrogen, Zn-Cl₂ (1.0 M solution in THF, 10 mL) was added to a cooled (0 °C) and stirred solution of epoxide **6** (2.16 g, 5.0 mmol) and glucal **3** (2.91 g, 7.5 mmol) in THF (25 mL). After 15 min, the reaction mixture was diluted with EtOAc (100 mL) and washed with sat. aq. NaCl (2 × 50 mL) and aq. NaHCO₃ (1.0 M, 50 mL). The organic phase was dried (MgSO₄) and concentrated in vacuo. Purification of the residue by silica gel column chromatography (20–50% EtOAc/light petroleum) afforded dimer glucal **30** (2.38 g, 2.9 mmol, 58%) contaminated with ca. 0.5 mmol (0.4 g) of the corresponding β -(1→4)-linked disaccharide glucal.

 $\begin{array}{l} \beta \cdot (1 \rightarrow 3) \text{-linked disaccharide glucal:} \ ^{13}\text{C}\{^1\text{H}\} \ \text{NMR} \ (\text{CDCl}_3): \ \delta = 145.8 \\ (\text{C1}), \ 144.2 \ (\text{C}_q \ \text{Tr}), \ 138.6, \ 137.8, \ 137.6 \ (\text{C}_q \ \text{Bn}), \ 128.9 - 127.0 \ (\text{C}_{arom}), \ 102.9 \\ (\text{C1}', \ J_{\text{C},\text{H}} = 158.9 \ \text{Hz}), \ 100.2 \ (\text{C2}), \ 86.6 \ (\text{C}_q \ \text{Tr}), \ 84.4, \ 81.9, \ 77.8, \ 77.7, \ 74.7, \\ 74.6, \ 68.2 \ (\text{C3}-\text{C5}, \ \text{C2}' - \text{C5}'), \ 75.3, \ 75.1, \ 73.5 \ (\text{CH}_2 \ \text{Bn}), \ 69.0 \ (\text{C6}'), \ 63.1 \\ (\text{C6}); \ \text{MS} \ (\text{ESI}): \ m/z = 821 \ (M + \text{H}^+), \ 838 \ (M + \text{NH}_4^+), \ 843 \ (M + \text{Na}^+). \\ \beta \cdot (1 \rightarrow 4) \text{-linked disaccharide glucal:} \ \ ^{13}\text{C}\{^1\text{H}\} \ \text{NMR} \ (\text{CDCl}_3): \ \delta = 145.8 \\ (\text{C1}), \ 144.2 \ (\text{C}_q \ \text{Tr}), \ 138.7, \ 138.3, \ 137.6 \ (\text{C}_q \ \text{Bn}), \ 128.9 - 127.0 \ (\text{C}_{arom}), \ 103.8 \\ (\text{C1}', \ J_{\text{C},\text{H}} = 161.2 \ \text{Hz}), \ 100.1 \ (\text{C2}), \ 86.7 \ (\text{C}_q \ \text{Tr}), \ 84.3, \ 82.2, \ 79.0, \ 77.8, \ 74.8, \\ 74.5, \ 68.2 \ (\text{C3}-\text{C5}, \ \text{C2}'-\text{C5}'), \ 75.4, \ 75.3, \ 73.7 \ (\text{CH}_2 \ \text{Bn}), \ 68.8 \ (\text{C6}'), \ 63.2 \\ (\text{C6}). \end{array}$

1,5-Anhydro-4-O-benzyl-3-O-(2,3,4,6-tetra-O-benzyl-β-D-glucopyranosyl)-6-O-trityl-2-deoxy-D-arabino-hex-1-enitol (31): NaH (0.24 g, 10 mmol) and benzyl bromide (1.2 mL, 10 mmol) were added to a stirred solution of dimer **30** (2.38 g, 2.9 mmol of β -(1 \rightarrow 3)-linked disaccharide and 0.4 g, 0.5 mmol of β -(1 \rightarrow 4)-linked dimer) in DMF (10 mL). The reaction mixture was quenched after 2 h by addition of MeOH (2 mL) and concentrated under reduced pressure. The residue was then dissolved in diethyl ether (50 mL), washed with H_2O (2×25 mL), dried (MgSO₄) and purified by silica gel column chromatography (0-20% EtOAc/light petroleum) to furnish fully protected dimer glucal 31 (2.52 g, 2.5 mmol, 87%) as a colourless oil. ¹H NMR (CD-Cl₃): $\delta = 7.50 - 7.05$ (m, 40 H, H_{arom}), 6.49 (dd, 1 H, H1, $J_{1,2} = 6.1$ Hz, $J_{1,3} = 0.6$ Hz), 4.87 (dd, 1 H, H2, $J_{2,3} = 3.0$ Hz), 4.82 (AB, 2 H, CH₂ Bn), 4.77 (AB, 2H, CH₂ Bn), 4.60 (AB, 2H, CH₂ Bn), 4.58 (AB, 2H, CH₂ Bn), 4.52 (AB, 2H, CH₂ Bn), 4.47 (d, 1H, H1', $J_{1,2} = 7.8$ Hz), 4.39 (ddd, 1H, H3, $J_{3,4} = 7.8$ Hz), 4.11 (m, 1H, H5), 3.95 (dd, 1H, H4, $J_{4,5} = 5.5$ Hz), 3.65-3.54 (m, 4H, H3'/H4'/H5'/H6A), 3.46 (m, 2H, H6'A/H6'B), 3.38 (m, 1H, H6B), 3.35 (dd, 1 H, H2', $J_{2',3'} = 8.7$ Hz); ¹³C{¹H} NMR (CDCl₃): $\delta = 144.6$ (C1), 143.7 (C_q Tr), 138.4, 138.2, 138.1, 138.0, 137.9 (C_q Bn), 128.5–126.7 (C_{arom}), 101.8 (C1'), 99.0 (C2), 86.2 (C_{q} Tr), 84.4, 81.9, 77.5, 76.6, 74.5, 74.2, 74.2 (C3-C5, C2'-C5'), 75.2, 74.6, 74.4, 73.0, 72.6 (CH₂ Bn), 68.7 (C6'), 62.3 (C6); C₆₆H₆₄O₉ (1000.5): calcd. C 79.18, H 6.44; found C 79.16, H 6.46.

1,2-Anhydro-4-*O*-benzyl-3-*O*-(**2,3,4,6-tetra-***O*-benzyl-β-D-glucopyranosyl)-6-*O*-trityl-β-D-glucopyranose (**32**): A solution of 3,3-dimethyldioxirane in acetone (0.080 m, 38 mL, 3.0 mmol) was added dropwise to a cooled (0 °C) and stirred solution of dimer glucal **31** (2.52 g, 2.5 mmol) in CH₂Cl₂ (10 mL). After the reaction mixture had been stirred for 5 min, the solvents were evaporated in vacuo to give 1,2-anhydro derivative **32** as a white foam (2.41 g, 2.4 mmol, 95%). ¹H NMR (CDCl₃): $\delta = 7.51 - 6.85$ (m, 40 H, H_{arom}), 5.09 (d, 1 H, H1, $J_{1,2} = 2.4$ Hz), 4.79 (AB, 2H, CH₂ Bn), 4.71 (AB, 2H, CH₂ Bn). 4.63 (AB, 2H, CH₂ Bn), 4.59 (d, 1 H, H1', $J_{1',2'} = 7.8$ Hz), 4.58 (AB, 2 H, CH₂ Bn), 4.52 (AB, 2 H, CH₂ Bn), 4.19 (dd, 1 H, H3, $J_{2,3} = 1.5$ Hz, $J_{3,4} = 7.5$ Hz), 3.95 (dd, 1 H, H4, $J_{4,5} = 6.5$ Hz), 3.72 (m, 1 H, H5), 3.70 (.352 (m, 4H, H3'/H4'/H5'/H6A), 3.41 (m, 2 H, H6'A/H6'B), 3.31 (m, 1 H, H6B), 3.29 (dd, 1 H, H2', $J_{2',3'} = 8.0$ Hz), 3.07 (dd, 1 H, H2); ¹³C(¹H} NMR (CDCl₃): $\delta = 143.6$ (C_a Tr), 138.3, 138.1, 138.0, 137.8, 137.7 (C_a Bn), 128.4 - 126.6 (C_{arom}), 101.0 (C1'), 86.0 (C_a Tr), 84.5, 81.9, 81.8, 77.4, 76.6, 72.8, 72.7, 68.9 (C1/C3 - C5, C2'-C5'), 75.2, 74.8, 74.5, 73.1, 73.0 (CH₂ Bn), 68.4 (C6'), 61.8 (C6), 51.8 (C2).

2-O-Acetyl-4-O-benzyl-3-O-(2,3,4,6-tetra-O-benzyl-B-D-glucopyranosyl)-6-Otrityl-β-D-glucopyranosylamine (33): Under a continuous stream of dry nitrogen, ZnCl₂ (1.0 M solution in THF, 3.6 mL) was added to a stirred solution of epoxide 32 (2.41 g, 2.4 mmol) in CH₃CN (20 mL). The reaction mixture was stirred for 2 h, subsequently diluted with EtOAc (100 mL) and washed with sat. aq. NaCl $(2 \times 50 \text{ mL})$ and aq. NaHCO₃ (1.0 M, 50 mL). The organic phase was dried (MgSO4) and concentrated in vacuo. To a stirred solution of residue in THF/H₂O (15 mL, 5:1, v/v) was added aq. H₂SO₄ (1.0 M, 0.24 mL). After 15 min, the reaction mixture was quenched by addition of aq. NaHCO₃ (1.0 m, 10 mL). The neutralised mixture was extracted with EtOAc $(2 \times 50 \text{ mL})$ and the combined organic layers were dried (MgSO₄) and concentrated under reduced pressure. The resulting colourless oil was purified by flash chromatography over silica gel (20-50% EtOAc/light petroleum) to furnish glucosylamine 33 as a white solid (1.52 g, 1.4 mmol, 59%). $[\alpha]_{\rm D} = +13.0^{\circ} (c = 0.2, \text{CHCl}_3); {}^{1}\text{H} \text{NMR} (\text{CDCl}_3): \delta = 7.55 - 6.90 \text{ (m, 40 H,}$ H_{arom}), 4.91 (dd, 1 H, H2, $J_{1,2} = 9.8$ Hz, $J_{2,3} = 8.9$ Hz), 4.81 (AB, 2 H, CH₂ Bn), 4.72 (AB, 2H, CH₂ Bn), 4.67 (AB, 2H, CH₂ Bn), 4.60 (AB, 2H, CH₂ Bn), 4.57 (AB, 2H, CH₂ Bn), 4.52 (d, 1H, H1', $J_{1', 2'} = 10.8$ Hz), 4.04 (d, 1H, H1), 3.73 (dd, 1 H, H6A, $J_{5, 6A} = 1.7$ Hz, $J_{6A, 6B} = 9.2$ Hz), 3.62 (dd, 1 H, H3', $J_{2', 3'} = 8.9$ Hz, $J_{3', 4'} = 7.8$ Hz), 3.59–3.45 (m, 7 H, H3/H4/H5/H4'/H5'/ H6'A/H6'B), 3.42 (dd, 1 H, H2'), 3.19 (dd, 1 H, H6B, $J_{5,6B} = 4.6$ Hz), 2.09 (s, 3H, CH₃ Ac), 1.61 (brs, 2H, NH₂); ${}^{13}C{}^{1}H{}$ NMR (CDCl₃): $\delta = 170.1$ (C=O Ac), 143.8 (C_q Tr), 138.3, 138.2, 138.1, 137.9, 137.8 (C_q Bn), 128.6 -126.7 (C_{arom}), 103.3 (C1'), 86.2 (C_q Tr), 84.7 (C1), 84.4, 81.9, 80.3, 77.9, 77.8, 76.4, 75.6, 75.0 (C2 - C5, C2' - C5'), 75.3, 74.6, 74.4, 73.1, 73.0 (CH, Bn), 69.0 (C6'), 63.1 (C6), 21.0 (CH₃ Ac); MS (ESI): $m/z = 1076 (M + H^+)$, 1093 $(M + NH_4^+)$; C₆₈H₆₉NO₁₁ (1075.5): calcd. C 75.89, H 6.46, N 1.30; found C 75.80, H 6.60, N 1.27.

Methyl 6-*N*-[2-*O*-acetyl-3-*O*-(2,3,4,6-tetra-*O*-benzyl-β-D-glucopyranosyl)-4-*O*-benzyl-β-D-glucopyranosyl]-2,3,4-tri-*O*-benzyl-α-D-amidoglucuronopyranoside (35): Yield: 0.934 g, 0.72 mmol, 91 %; $^{-13}C{}^{1}H{}$ NMR (CDCl₃): $\delta = 170.6, 169.1$ (C6A, C=O Ac), 138.1 – 137.5 (C_q Bn), 127.9 – 127.0 (C_{arom}), 103.2 (C1C), 98.0 (C1A), 84.5 (C1B), 81.8, 81.6, 80.7, 79.8, 79.7, 78.9, 77.3, 76.7, 74.9, 73.4, 70.2, 70.1 (C2A – C5A, C2B – C5B, C2C – C5C), 75.3 – 72.9 (CH₂ Bn), 68.7 (C6C), 61.0 (C6B), 55.2 (OMe), 20.5 (CH₃ Ac); C₇₇H₈₃NO₁₇ (1293.6): calcd. C 71.44, H 6.46, N 1.08; found C 71.60, H 6.61, N 1.02.

Methyl 6-*N*-[2-*O*-acetyl-3-*O*-(2,3,4,6-tetra-*O*-benzyl-β-D-glucopyranosyl)-4-*O*-benzyl-β-D-glucuronopyranosyl]-2,3,4-tri-*O*-benzyl-α-D-amidoglucuronopyranoside (36): Yield: 0.734 g, 0.56 mmol, 78%; ${}^{13}C{}^{1}H{}$ NMR (CDCl₃): $\delta = 170.8, 170.2, 169.9$ (C6A/C6B, C=O Ac), 138.3–137.4 (C_q Bn), 128.2– 127.3 (C_{arom}), 103.4 (C1C), 98.2 (C1A), 84.7 (C1B), 82.0, 81.1, 81.0, 79.8, 79.7, 79.1, 79.0, 77.8, 77.7, 76.8, 73.4, 70.1 (C2A–C5A, C2B–C5B, C2C– C5C), 75.4–73.2 (CH, Bn), 68.9 (C6C), 55.6 (OMe), 20.6 (CH₃ Ac); $\rm C_{77}H_{81}NO_{18}$ (1308.5): caled. C 70.68, H 6.24, N 1.07; found C 70.59, H 6.30, N 1.10.

Methyl 6-N-[6-N-[2-O-acetyl-3,4-di-O-benzyl-6-O-trityl-β-D-glucopyranosyl]-2-O-acetyl-3-O-(2,3,4,6-tetra-O-benzyl-β-D-glucopyranosyl)-4-O-benzyl-β-D-amidoglucuronopyranosyl]-2,3,4-tri-O-benzyl-α-D-amidoglucuronopyranoside (37): Yield: 0.881 g, 0.45 mmol, 81%; $^{13}C{}^{1H}$ NMR (CDCl₃): δ = 170.4, 170.2, 169.3, 167.6 (C6A/C6B, 2C=O Ac), 143.4 (C_q Tr), 138.1–137.2 (C_q Bn), 127.9–126.7 (C_{arom}), 103.2 (C1C), 97.6 (C1A), 85.9 (C_q Tr), 84.5, 82.7 (C1B/C1D), 80.7–69.9 (C2A–C5A, C2B–C5B, C2C–C5C, C2D–C5D), 75.1–72.9 (CH₂ Bn), 68.7 (C6C), 61.5 (C6D), 54.5 (OMe), 20.4, 20.3 (CH₃ Ac); MS (ESI): m/2z = 968 (M+2H⁺); C₁₁₈H₁₂₀N₂O₂₃ (1932.8): calcd. C 73.27, H 6.25, N 1.45; found C 73.12, H 6.40, N 1.39.

$\label{eq:methyl} Methyl = 6-N-[6-N-[2-O-acetyl-3,4-di-O-benzyl-\beta-D-glucopyranosyl]-2-O-acetyl-3-O-(2,3,4,6-tetra-O-benzyl-\beta-D-glucopyranosyl)-4-O-benzyl-\beta-D-amido-tyl-\beta-D-amido-tyl-\beta-D-glucopyranosyl)-4-O-benzyl-\beta-D-amido-tyl-\beta-D-glucopyranosyl)-4-O-benzyl-\beta-D-amido-tyl-\beta-D-glucopyranosyl)-4-O-benzyl-\beta-D-amido-tyl-\beta-D-glucopyranosyl)-4-O-benzyl-glucopyranosyl-glucopy$

algorithmorpy ratiosy ($_{23}$), $_{41}$ ($_{13}$), $_{13}$ ($_{14}$) MMR (CDCl₃): $\delta = 170.5$, 170.2, 169.9, 169.7, 168.4 (C6A/C6B/C6D, 2C=O Ac), 138.2–137.3 (C_q Bn), 128.5–127.1 (C_{aron}), 103.3 (C1C), 98.0 (C1A), 84.5, 81.9 (C1B/C1D), 80.8– 70.0 (C2A–C5A, C2B–C5B, C2C–C5C, C2D–C5D), 75.5–73.1 (CH₂ Bn), 68.8 (C6C), 55.4 (OMe), 20.5, 20.4 (CH₃ Ac); C₉₉H₁₀₄N₂O₂₄ (1704.7): calcd. C 69.70, H 6.14, N 1.64; found C 69.54, H 6.28, N 1.57.

Methyl 6-*N*-[6-*N*-[2-*O*-acetyl-3-*O*-(2,3,4,6-tetra-*O*-benzyl-β-D-glucopyranosyl)-4-*O*-benzyl-β-D-glucuronopyranosyl]-2-*O*-acetyl-3,4-di-*O*-benzyl-β-Damidoglucuronopyranosyl]-2-*O*-acetyl-3-*O*-(2,3,4,6-tetra-*O*-benzyl-β-D-glucopyranosyl)-4-*O*-benzyl-β-D-amidoglucuronopyranosyl]-2,3,4-tri-*O*-benzyl-*x*-Damidoglucuronopyranoside (42): Yield: 0.376 g, 0.15 mmol, 74%; ¹³C{¹H} NMR (CDCl₃): $\delta = 171.7$, 170.6, 170.2, 169.6, 169.5, 169.4, 167.9 (C6A/C6B/ C6D/C6B', 3C=O Ac), 138.3–137.0 (C_q Bn), 128.1–127.2 (C_{arom}), 103.3, 103.2 (C1C/C1C'), 98.1 (C1A), 84.6, 81.9, 81.1 (C1B/C1D/C1B'), 80.9–70.2 (C2A - C5A, C2B - C5B, C2C - C5C, C2D - C5D, C2B' - C5B', C2C' - C5C'), 75.6 73.1 (CH₂ Bn), 68.9, 68.8 (C6C/C6C'), 55.5 (OMe), 20.7, 20.6, 20.4 (CH₃ Ac); C₁₄₈H₁₅₅N₃O₃₅ (2534.0): calcd. C 70.10, H 6.16, N 1.66; found C 70.14, H 6.28, N 1.59. Methyl 6-N-[6-N-[6-N-[2-O-acetyl-3,4-di-O-benzyl-6-O-trityl-B-D-glucopyranosyl]-2-O-acetyl-3-O-(2,3,4,6-tetra-O-benzyl-\$-D-glucopyranosyl)-4-O-benzyl-β-D-amidoglucuronopyranosyl]-2-O-acetyl-3,4-di-O-benzyl-β-D-amidoglucuronopyranosyl]-2-O-acetyl-3-O-(2,3,4,6-tetra-O-benzyl-β-D-glucopyranosyl)-4-O-benzyl-\$-D-amidoglucuronopyranosyl]-2,3,4-tri-O-benzyl-\$\alpha-D-amidoglucuronopyranoside (43); Yield: 0.357 g, 0.12 mmol, 77%; ¹³C{¹H} NMR $(CDCl_3)$: $\delta = 171.6, 170.8, 170.6, 170.4, 170.3, 169.6, 168.5, 167.8 (C6A/C6B/$ C6D/C6B', 4C=OAc), 143.7 (C_q Tr), 138.4, 138.3, 138.3, 138.2, 138.0, 138.0, 137.9, 137.9, 137.8, 137.8, 137.7, 137.7, 137.6, 137.6, 137.5, 137.4, 137.3 (C_a Bn), 127.9-126.9 (C_{arom}), 103.4, 103.4 (C1C/C1C'), 98.2 (C1A), 86.3 (C_a Tr), 84.7, 84.6, 83.5, 83.1 (C1B/C1D/C1B'/C1E), 82.0, 81.9, 81.4, 81.1, 80.2, 79.4, 79.3, 79.2, 78.2, 78.1, 77.9, 77.7, 77.3, 77.2, 76.8, 76.0, 75.8, 74.5, 74.5, 74.3, 73.6, 73.4, 73.2, 72.9, 72.8, 72.6, 71.7, 70.3 (C2A-C5A, C2B-C5B, C2C-C5C, C2D--C5D, C2B'-C5B', C2C'-C5C'/C2E-C5E), 75.8, 75.8, 75.7, 75.6, 75.6, 75.4, 75.4, 75.4, 75.3, 75.3, 75.2, 75.1, 75.0, 75.0, 74.8, 74.8, 74.5 (CH2 Bn), 69.0, 68.9 (C6C/C6C'), 62.4 (C6E), 55.7 (OMe), 20.9, 20.7, 20.6, 20.5 (CH₃ Ac); MS (ESI): $m/2z = 1581 (M + 2H^+)$; $C_{189}H_{194}N_4O_{40}$ (3159.3): caled. C 71.80, H 6.18, N 1.77; found C 71.99, H 6.40, N 1.65.

$\label{eq:started_st$

(2): Compound 43 (357 mg, 0.12 mmol) was deprotected and purified as described in the general procedure to give 2 as a white fluffy solid (98 mg, 0.08 mmol, 70%). NMR data are given in Table 2. $[\alpha]_D = -2.0^\circ$ (c = 0.3, H₂O); MS (ESI): m/2z = 610 (M + 2H⁺); C₄₃H₇₀N₄O₃₆ (1218.4): calcd. C 42.37, H 5.79, N 4.60; found C 42.31, H 5.93, N 4.53.

Table 2. ${}^{13}C{}^{1}H$ NMR (D₂O, 150.3 MHz) and ${}^{1}H$ NMR (D₂O, 600 MHz) data for heptamer 2 [a] obtained from CH-COSY, HH-COSY and TOCSY experiments.

	1	2	3	4	5	6	6'
A	4.87 [b] (d) [c] (3.8) [d]	3.62 (dd) (3.8, 9.8)	3.69 (dd) (9.8, 10.0)	3.55 (dd) (10.0, 9.8)	4.09 (d) (9.8)		
в	100.5 [e] 5.15 (d) (9.3)	71.5 3.74 (t) (9.2)	70.5 3.89 (dd) (9.2, 9.9)	78.5 3.69 (t) (9.9)	77.2 4.08 (d) (9.9)	[f]	
B′	80.0 5.12 (d) (9.3)	71.4 3.74 (t) (9.3)	84.6 3.89 (dd) (9.3, 9.9)	77.5 3.69 (t) (9.9)	72.1 4.05 (d) (9.9)	[f]	
С	80.0 4.77 (d)	71.4 3.34 (dd)	84.5 3.50 (dd)	77.5 3.39 (t)	72.0 3.46 (m)	[f] 3.91 (dd) (12 3 2 3)	3.71 (dd)
C'	(7.9) 103.5 4.76 (d)	(7.9, 6.9) 74.3 3.35 (dd) (8.0, 8.9)	(8.9, 8.6) 76.4 3.51 (dd)	70.1 3.38 (t)	77.3 3.46 (m)	61.6 3.90 (dd) (12.3, 2.3)	3.70 (dd) (12 3 4 1)
D	(0.0) 103.5 5.10 (d)	(0.0, 0.9) 74.3 3.51 (dd)	(8.9, 6.0) 76.4 3.60 (t)	(8.0) 70.1 3.59 (t)	77.3 4.03 (d)	61.6	(12.3, 4.1)
E	79.9 5.01 (d) (9.2)	().3, 7.2) 71.9 3.36 (t) (9.2)	(9.2) 77.2 3.43 (t) (9.2)	(9.3) 77.4 3.41 (dd) (9.2, 8.9)	71.8 3.51 (m)	[f] 3.85 (dd) (11.6, 2.5)	3.68 (dd) (11.6, 5.0)
	79.8	73.4	72.5	71.3	77.0	61.4	

[a] OMe group: $\delta = 3.61$ (¹H NMR) and 56.4 (¹³C NMR). [b] Proton chemical shift. [c] Proton multiplicity. [d] Proton coupling constants (Hz). [e] ¹³C chemical shift. [f] C6A³ C6B/C6D/C6B': 173.3, 172.4, 172.1, 171.9.

Acknowledgements: The work described in this paper was supported by the Netherlands Foundation for Chemical Research (SON) with financial aid from the Netherlands Organization for Scientific Research (NWO). We thank Dr. E. M. Mösinger (Sandoz Agro, Switzerland) for testing compound **2** for phytoalexin-elicitor activity.

Received: January 17, 1997 [F 578]

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